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## $2\pi$ + $8\pi$ CYCLOADDITION REACTIONS OF 3-OXIDOPYRIDINIUM BETAINES<sup>1</sup>

Nicholas Dennis, Alan R. Katritzky\* and Gebran J. Sabounji

School of Chemical Sciences, University of East Anglia, Norwich, NR4 7TJ

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We have reported cycloadditions of 3-oxidopyridinium betaines <u>1</u> in which they act as  $4\pi$ -components (adding  $2\pi^2$  or  $6\pi^3$  addends across the 2, 6-positions) or as  $2\pi/6\pi$ components (adding thermally  $4\pi^4$  or photochemically  $2\pi/6\pi^5$  addends across the 2, 4positions). We now describe cycloadditions in which they act as  $8\pi$  components, adding  $2\pi$  addends across the oxygen and the carbon 4.

Dichloroketene (2, R' = C1) generated <u>in situ</u> from dichloroacetyl chloride<sup>6</sup> or from chloral,<sup>7</sup> in the presence of betaines <u>la-d</u><sup>8</sup> gave, we believe <u>via</u> intermediates <u>3</u> which spontaneously lost HC1, the new bicyclic compounds <u>4a-d</u> (Table).<sup>9</sup> Monochloroketene (2, R' = H) generated from monochloroacetyl chloride<sup>10</sup> yielded compound <u>4e</u> (R' = H) (Table). For example, treatment of 3-hydroxy-1-phenylpyridinium chloride<sup>11</sup>, the precursor of betaine <u>1b</u><sup>12</sup> and dichloroacetyl chloride in CH<sub>2</sub>Cl<sub>2</sub> with NEt<sub>3</sub> at 0°C gave the adduct <u>3H-7-oxo-3-phenylfuro[3, 2-d]</u> pyridine, <u>4b</u>. The structures <u>4</u> are supported by UV, NMR and mass spectra; the IR spectra (Table) are characteristic for  $\alpha\beta$ -unsaturated  $\gamma$ -lactones.<sup>13</sup> The NMR spectrum had a 1H doublet at  $\delta$  6.90 (<u>J</u> = 8 Hz), a <u>5H</u> multiplet at  $\delta$  7.50, a 1H doublet of doublets at  $\delta$  8.10 (<u>J</u> = 8 and 2 Hz) and a 1H fine doublet at  $\delta$  8.30 (<u>J</u> = 2 Hz) assigned to the olefinic H-5, the phenyl, the olefinic H-4 and H-2 protons of <u>4c</u> respectively.



 $R = \underline{a}$ ,  $CH_2Ph$ ;  $\underline{b}$ , Ph;  $\underline{c}$ , 3, 5-dimethoxy-2, 4, 6-triazin-1-yl;  $\underline{d}$ , 3, 5-dimethyl-2, 6pyrimidin-1-yl;  $\underline{e}$ , Ph. R' = Cl for  $\underline{a} - \underline{d}$ ; R' = H for  $\underline{e}$ . Ketenes generally undergo concerted  $\pi 2s + \pi 2a$  cycloadditions, <sup>14</sup> although formal  $2\pi + 4\pi$  additions of ketenes have been reported. <sup>15</sup> Recently, dichloroketene (2, R' = Cl) was reacted <sup>16</sup> with tropone in a  $2\pi + 8\pi$  sense to yield a bicyclic lactone by elimination of HCl from an intermediate dichlorolactone.

TABLE. PHYSICAL PROPERTIES OF 3-SUBSTITUTED-7-OXO-3H-FURO[3,2-d]PYRIDINES

Cpd.	$IR(\nu_{max})(CHBr_3)(film)$				$\lambda_{\text{max}} (\log \epsilon) (CH_3 CN)$	Crystal form, m.p. ( <sup>o</sup> C)
	C =O	C≂C	C=N	N-C <b>=</b> C		solvent, yield (%)
<u>4a</u>	1740	1545			408 (4.47), 396s (4.37), 265 (3.80)	gold flakes, 279–280 <sup>0</sup> (EtOH)(25%)
<u>4b</u>	1730	1660			366 (4.70), 235 (3.00)	yellow needles, 222-223 <sup>0</sup> (EtOH) (57%)
<u>4c</u>	1734		1600	1670	398 (4.34), 378 (4.36), 217 (3.56)	red needles 245-246 <sup>0</sup> , (DMSO) (30%)
<u>4d</u>	1740	1610		1660	396 (4.46), 374 (4.06), 237 (3.08)	orange-red needles, 275- 276 <sup>0</sup> (EtOH) (85%)
<u>4e</u>	1740	$1660 \\ 1640$			367 (4.60), 245 (3.00)	needles 220-222 <sup>0</sup> (dec) (MeOH/EtOEt) (25%)

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